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CHAPTER 8 SOLUTIONS

MANUAL Covalent

Bonding Covalent

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Matter and Change •

Chapter 8 121 Section

8.1 The Covalent Bond

pages 240–247

Practice Problems page

244 Draw the Lewis

structure for each

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## PDF Chapter 8

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molecule. 1.  $\text{PH}_3$   $\text{H}_2\text{O}$   $\text{H}_2$   
 $\text{H}-\text{H}$   $\text{H}-\text{H}$   $\text{P}=\text{O}$  respectively,  
for single, double, and  
triple  $\text{P}\equiv\text{P}$  — — 2.  $\text{H}_2$   $\text{S}_2$   $\text{H}_2$   
 $\text{H}-\text{H}$  —  $\text{H}-\text{S}-\text{S}-\dots$

### **Covalent Bonding Covalent Bonding**

Section 8.2 - The  
Nature of Covalent  
Bonding In ionic  
bonding, atoms  
transfer electrons to  
achieve noble gas  
configuration. In  
covalent bonding,

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## PDF Chapter 8

### Covalent Bonding

atoms share electrons to achieve noble gas configuration. Most atoms share electrons until they have a total of 8 valence electrons (octet rule).

## **Chapter 8 - Covalent Bonding**

Chapter 8 Covalent Bonding and Molecular Structure 8-3 There are two types of repulsive forces between the two atoms. First, the nuclei repel because they are

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both positively charged. Second, the electrons repel because they are both negatively charged.

**Chapter 8: Covalent Bonding and Molecular Structure**

Chapter 8 Covalent Bonding 181. Section Review. Objectives.

- Distinguish molecular compounds from ionic compounds.
- Identify the information a molecular formula

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provides. Vocabulary  
Part A Completion. Use  
this completion  
exercise to check your  
understanding of the  
concepts and terms  
that are introduced in  
this section.

**05 CTR ch08 7/12/04**  
**8:12 AM Page 181**  
**MOLECULAR**  
**COMPOUNDS 8**

Complete the Lewis  
Diagrams of Covalent  
Compounds worksheet.  
Complete the Terms



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Covalent Bonding  
and Concepts found in  
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Chapter 8 Sections 8.2

The Nature of Covalent  
Bonding Read the  
section Take notes in  
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**Term 2, Core Module  
6 Chapter 8,  
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Answers to Help Your  
Chemistry Problems.  
Chemistry is not the  
easiest lesson at high-  
school, and it is even  
considered as one of  
the hardest subjects.  
Chemical bonding  
might be one of the  
difficult chemistry  
subjects.

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**Ionic And Covalent**

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chemical bonds ionic.

**Covalent  
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242 Chapter 8 •  
Covalent Bonding  
Single Covalent Bonds  
When only one pair of  
electrons is shared,  
such as in a hydrogen  
molecule, it is a single  
covalent bond The  
shared electron pair is  
often referred to as the  
bonding pair For a  
hydrogen molecule,

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## PDF Chapter 8

### Covalent Bonding

shown in Figure 84,  
each covalently  
bonded atom equally  
attracts the pair of  
shared

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Start studying Chapter  
8 Covalent Bonding  
Sections 8.1 and 8.2.  
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flashcards, games, and  
other study tools.

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Worksheet

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Chapter 8 Covalent Bonding Sections 8.1 and 8.2 Flashcards ...

bond actually looks like! Chapter 9 -

Covalent Bonding I.

Covalent Bonding:

attractive force

produced as a result of shared (pgs

189electrons. -193) A.

A \_\_\_\_\_ is formed when

two or more atoms

bond covalently. B.

Bonds form when there

is a balance of

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PDF Chapter 8

Covalent Bonding

attractive and  
repulsive forces

between two atoms: C.  
Single Covalent Bond &  
Lewis Structures

## **Chapter 6 - Covalent Bonding**

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Key Worksheets -  
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Chemistry - Chapter 8 -

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### Covalent Bonding

Gravity. Created by.

PurpleButterflyLK.

These are the vocabulary and key questions for chapter 8 of the chemistry textbook. I hope you appreciate it as much as I do, since it took me 1 1/2 hours to fill out this information. =)

## **Chemistry - Chapter 8 - Covalent Bonding Flashcards | Quizlet**

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