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answer Molar Mass (# grams) Atomic Weight (for elements) Gram Formula/Molecular Weight (for compounds) 23 ATOMS (for elements) or 6.02 X 10²³ MOLECULES ... CHEMISTRY MOLES PACKET PAGE 8 — STP. = 2 = CHEMISTRY MOLES PACKET ...

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Chemistry B. Moles Packet INTRODUCTION TO MOLES. We are about to start on a unit of chemical calculations called "stoichiometry". Stoichiometry is how we calculate the relationships between the amounts of reactants and the amounts of products.

CHEMISTRY WORKSHEET # 2 MOLE PROBLEMS—THE MOLE AS A UNIT ...
CHEMISTRY B- MOLES PACKET NAME: _____ HR: _____ PAGE 3 CHEMISTRY WORKSHEET # 1 MOLAR MASS (GRAM MOLECULAR/FORMULA WEIGHTS) We know that grams are actually a measure of the mass of matter and not the weight. Mass is the quantity of matter present; weight is a measure of the pull of gravity on matter and is measured in pounds or newtons.

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A) 8.5 moles of sodium acetate dissolved in 1 L of water B) 5.5 moles of sodium acetate dissolved in 500 mL of water C) 1.8 moles of sodium acetate dissolved in 300 mL of water D) 1.2 moles of sodium acetate dissolved in 200 mL of water E) none of the above

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