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chemical reaction. The
process by which the
atoms of one or more
substances are
rearranged to form
different substances;
occurs when there is
change in temperature,
color, odor, and

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physical change.
reactant.

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Vocabulary ...

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Mole. STUDY. PLAY.

mole. the SI base unit
used to measure the
amount of a substance.

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mole. is defined as the number of carbon atoms in exactly 12g of pure carbon- 12.

Avogadro's number. the number

6.02×10^{23} in honor of the Italian physicist and lawyer Amedeo Avogadro.

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Explain how a mole is

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surface on galvanized

steel. Determine the

number of Zn atoms in

2.50 mol of Zn. 2.50

mol Zn

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pair 2. 5 3. dozen 4.
gross 5. 200 6. ream 7.
6,000,000,000 8. 0.5
mol 9. 6.02 $\times 10^{23}$ 10.
four moles 11. 6.02 $\times 10^{23}$
Cu atoms 12. 1 mol Cu

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12. 4 23 4 1 mol CH
6.02 10 molecules CH
13. 23 1 mol Xe 6.02
10 molecules Xe 14. 23
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Causes of Change.
Enthalpy. Temperature.
Gibb's Energy.
Calorimeter. the sum of
the internal energy of

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the system. a measure of how hot something is; the measure of the average.... energy in a system that is available for work (also known as f....

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mol Zn

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compound contains**

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**3.86 g of sulfur and
4.08** Answer Key

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A chemist isolated four
samples, A, B, C, and

D. She obtained the

following atomic

emission spectra of the

samples. 400 500 600

700 29 Nanometers 1.

Examine each sample's

atomic emission

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spectra.

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