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enstructional Fair, Inc. 5. 35 g of
 $\text{CuSO}_4 \cdot 5\text{H}_2\text{O}$ Determine the number of
grams in each of the quantities below. 1,
2.5 moles of NaCl 2. 0.50 moles of 3,

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1.70 moles of KMnO_4 , 0.25 moles of KCl , ST 5. 3.2 moles of $\text{CuSO}_4 \cdot 5\text{H}_2\text{O}$
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element symbol that applies. CLUE 1. 2.

|He SAI

A chemical equation is the symbol in
Chemistry that represents chemical
reaction with the help of chemical
formulas. It contains the chemical

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substances that are involved in the reaction. It contains reactants and products. The reactants are the elements that react with one and another in a chemical reaction, while the products are the elements ...

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solids and liquids do not affect
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