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Study Guide Heat For Physics

Heat and temperature are different physical quantities. When two objects with different temperatures are in contact with each other, heat flows from the hotter system to the colder one. Heat is a measure of energy. An increase or decrease in mechanical energy in a system always accompanies an equal decrease or increase of heat, respectively.

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In a nutshell, thermal physics is the study of heat. Heat energy, or thermal energy, is the energy of a substance or system in terms of the motion or vibrations of its molecules. The faster the...

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The greater the ratio between the high temperature and the low temperature, the closer the efficiency becomes to 1. Therefore, an extremely low lower temperature and an extremely hot higher temperature yield a near perfectly efficient engine.

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Physics Chapter 9 Heat and Temperature. temperature. internal energy. thermal equilibrium. heat. a measure of the average kinetic energy of the particles in a.... the energy of a substance due to both the random motions of it.... the state in which two bodies in physical contact with each ot....

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Mathematically, it reads $\Delta Q = \Delta U + \Delta W$, where ΔQ is the heat energy supplied to the system, ΔU is the change in the internal energy, and ΔW is the work done by the system against external forces. It must be emphasized that these quantities are defined in general terms.

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Heat is a form of energy which can be neither created nor destroyed but can change its form 2nd law of thermodynamics Every energy transfer or transformation increases the entropy (randomness) of the universe

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convection. friction. The movement of heat within a solid or from one solid to another.... Any material that allows heat to move through it easily. The way in which heat moves through liquids and gases. A force that occurs when one object rubs against another object.... conduction.

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